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UTILITY PATENT APPLICATION TRANSMITTAL

Address to: Box PATENT APPLICATION Commissioner for Patents United States Patent & Trademark Office Washington, D.C. 20231	Attorney Docket No.	WAS 0394 PUS
	Inventor(s) or Appln. Identifier	Philipp Müller, Frank Achenbach, and Georg Eberl

1. This application entitled SELF-ADHESIVE ADDITION-CROSSLINKING SILICONE COMPOSITIONS is:

- a. X A new application under 37 C.F.R. § 1.53(b).
- b. A continuation divisional or continuation-in-part application under 37 C.F.R. § 1.53(b) of prior application Serial No. / filed on , entitled .

Application elements and other attached papers:

2. X Specification (including Claims and Abstract) [Total Pages 30]
3. Drawings (informal / formal) [Total Sheets]
4. X Oath or Declaration
- a. X Newly-executed
- b. Copy from a prior application (37 C.F.R. § 1.63(d))
5. Incorporation By Reference: The entire disclosure of the prior application, from which a copy of the oath or declaration is supplied under Item 4b, is considered as being part of the disclosure of the accompanying application and is hereby incorporated by reference therein.
6. This application is filed by fewer than all the inventors named in the prior application, 37 C.F.R. § 1.53(d)(4).
- a. **DELETE** the following inventor(s) named in the prior nonprovisional application:

- b. The inventor(s) to be deleted are set forth on a separate sheet attached hereto.

CERTIFICATION UNDER 37 C.F.R. § 1.10

I hereby certify that this UTILITY PATENT APPLICATION TRANSMITTAL and the documents referred to as attached therein are being deposited on the below date with the United States Postal Service in an envelope as "Express Mail Post Office to Addressee" addressed to: Box Patent Application, Commissioner for Patents, United States Patent and Trademark Office, Washington, D.C. 20231 on:

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9. Fee Calculation:

FOR	NUMBER FILED	NUMBER EXTRA	RATE	CALCULATIONS
TOTAL CLAIMS (37 C.F.R. § 1.16(c))	10 -20 =	0	x 18.00	0
INDEPENDENT CLAIMS (37 C.F.R. § 1.16(b))	1 -3 =	0	x 80.00	0
MULTIPLE DEPENDENT CLAIMS (if applicable) (37 C.F.R. § 1.16(d))			270.00	0
BASIC FEE (37 C.F.R. § 1.16(a))				710.00
TOTAL OF ABOVE CALCULATIONS				710.00
REDUCTION BY 50% FOR FILING BY SMALL ENTITY (NOTE 37 C.F.R. §§ 1.9, 1.27, 1.28)				0
\$40.00 ASSIGNMENT RECORDAL FEE (if applicable)				40.00
TOTAL				750.00

10. X A check in the amount of \$ 750.00 is enclosed.
11. X The Commissioner is hereby authorized to credit overpayments or charge the following fees (or any deficiency therein) to Deposit Account No. 02-3978 :
- a. X Fees required under 37 C.F.R. § 1.16.
- b. X Fees required under 37 C.F.R. § 1.17.

12. Maintenance of Copendency of Prior Application

_____ A request for extension of time and the appropriate fee have been filed in the pending **prior** application (or are being filed in the prior application concurrently herewith) to extend the period for response until _____.

13. _____ An Information Disclosure Statement (IDS) is attached, along with the following indicated attachments thereto:

a. _____ Form PTO/SB/08 (_____ sheet(s))

b. _____ Copies of references cited

14. _____ Certified copy of priority document(s)

15. X Return Receipt Postcard

16. X Other: Priority is claimed to German Application No. 199 59 412.0

17. X An Assignment of the invention to Wacker-Chemie GmbH

a. X is attached.

b. _____ was recorded on _____ at Reel _____, Frame _____.

18. The power of attorney in the prior application is to:

Name of Attorney of Record

Reg. No.

_____ The power appears in the original papers in the prior application.

_____ The power does not appear in the original papers, but was filed on _____.

_____ A new power has been executed and is attached.


19. Correspondence Address: Please address all future communications to:

William G. Conger,
Brooks & Kushman P.C.
1000 Town Center, 22nd Floor
Southfield, MI 48075-1351
Telephone: 248-358-4400; Fax: 248-358-3351

Respectfully submitted,

**PHILIPP MÜLLER, FRANK ACHENBACH,
AND GEORG EBERL**

Date October 13, 2000



Name: William G. Conger
Registration No.: 31,209

X Attorney or agent of record

Filed under Rule 34(a)

SELF-ADHESIVE ADDITION-CROSSLINKING SILICONE COMPOSITIONS

TECHNICAL FIELD

5 The invention relates to self-adhesive addition-crosslinking silicone compositions and addition-crosslinked silicone elastomers and composite materials prepared therefrom.

BACKGROUND ART

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10 It is known that the adhesion of addition-crosslinked silicone elastomers to numerous substrates, such as plastics, metals and glasses, is poor, i.e. if an addition-crosslinking silicone elastomer material is applied to a substrate and then crosslinked, the silicone elastomer formed can, as a rule, be peeled away from the substrate surface without problems. Only small tensile forces; frequently, even spontaneous delamination of the silicone elastomer from the substrate is found. However, since strong and permanent adhesion of the silicone elastomer to the
15 substrate is of primary importance in numerous applications, a large number of special measures have been proposed for achieving a strong bond between substrate and silicone elastomer.

20 In principle, the adhesive strength of the silicone elastomer/substrate composite can be increased by suitably changing the chemical and/or physical characteristics of the substrate or its surface prior to application of the addition-crosslinking silicone elastomer composition. This can be effected, for example, by pretreating the substrate surface with adhesion-promoting additives (so-called primers), by subjecting the substrate surface to a plasma treatment, by formulating the substrate to contain special additives, by selectively adjusting the morphology of the substrate,
25 by increasing the surface roughness, etc. These measures have, inter alia, the disadvantage that additional process steps are required. As the characteristics of the substrate often have to meet special requirements, use of these methods of increasing adhesive strength is often not possible.

EP-A-686 671 describes a self-adhesive adhesion-crosslinking material which employs no special adhesion promoter, because the adhesion-promoting component is either an organohydrogenpolysiloxane which has on average at least two SiH groups per molecule and whose monovalent Si-bonded radicals comprise at least 12 mol% of hydrocarbon radicals having an aromatic ring; or is a compound which has on average at least one SiH group per molecule and contains a group consisting of two aromatic rings, the two aromatic rings being separated from one another by $-R^{13}R^{14}Si-$, $-R^{13}R^{14}SiO-$, $-OR^{13}R^{14}SiO-$ or $-R^{13}R^{14}SiOR^{13}R^{14}Si-$, the radicals R^{13} and R^{14} being monovalent hydrocarbon radicals. The adhesion-promoting component can thus simultaneously function as the crosslinking agent of the silicone elastomer material. Good adhesion to organic plastics (especially ABS) is achieved with this composition, while at the same time the cured or partially cured products exhibit good demoldability from the metallic vulcanization mold (chromium- or nickel-coated steel mold or mold of an aluminum alloy). The high content of greater than 12 mol% of radicals containing aromatic rings in the SiH-containing, adhesion-promoting component results, however, in considerable incompatibility with the

other components of the addition-crosslinking silicone elastomer material. This leads to partial separation (exudation) during storage, necessitating repeated homogenization of the ingredient containing this component before use. This incompatibility, which is already evident from a milky turbidity of the uncrosslinked material, also manifests itself in substantially reduced transparency of the silicone elastomer parts produced therefrom. If the adhesion-promoting component simultaneously acts as a crosslinking agent of the silicone elastomer composition, the incompatibility leads to vulcanization problems, which result in inhomogeneous network formation and poor mechanical vulcanization properties. To overcome these vulcanization problems, it is necessary, in addition to the adhesion-promoting SiH-containing component, to use an SiH-containing crosslinking agent completely compatible with the silicone elastomer material, which however results in other disadvantages, for example higher values of the compression set, and higher tendency to exudation of the adhesion-promoting component. The high content of greater than 12 mol% of radicals containing aromatic rings in the SiH-containing, adhesion-promoting component also results in a considerable structural viscosity and thixotropy of the silicone elastomer material, which is undesired in numerous applications, for example, injection molding of liquid silicone rubber. Finally, the adhesion of this composition to metals is also insufficient.

EP-A-875 536 describes a self-adhesive adhesion-crosslinking silicone rubber mixture which is distinguished by the fact that

- a) the SiH crosslinking agent contains at least 20 SiH groups, other radicals being aliphatically saturated,
- b) an alkoxysilane and/or alkoxysiloxane having epoxy functional groups is/are present, and
- c) a peroxide is optionally present.

The use of glycidyloxypropyltrimethoxysilane (Glymo) is particularly preferred. The silicone rubber mixture described in EP-A-875 536 is particularly suitable for the production of composite shaped articles which consist of the silicone elastomer and an organic plastic. However, the composition described in EP-A-875 536 has the disadvantage that sufficient adhesive strength can be achieved only with the use of

very SiH-rich crosslinking agents having on average at least 20 SiH groups per molecule. In the examples there, crosslinking agents having 30 SiH groups per molecule are used. The use of such polyfunctional crosslinking agents considerably reduces the shelf-life of addition-crosslinking silicone rubber mixtures, i.e. the flowability is considerably impaired, which may lead to stiffening of the material. As a result, proper processing of the material, for example by injection molding, is no longer possible. In addition, in order to achieve high adhesive strength, it is necessary to use relatively large amounts of alkoxysilane/alkoxysiloxane having epoxy functional groups, with the result that the crosslinking rate is considerably reduced. Although this can be partly compensated by using a peroxide, as described in EP-A-875 536, only peroxides having a low initiation temperature, such as the 2,4-dichlorobenzoyl peroxide described, are suitable for this purpose, due to the necessarily low crosslinking temperature (softening of the organic plastic). These peroxides on the one hand are toxicologically very unsafe owing to the cleavage products and secondary products liberated (PCB problem) and on the other hand further impair the shelf-life of the material.

In summary, it may be said that none of the conventional addition-crosslinking silicone elastomer compositions satisfactorily meet the requirements set for a self-adhesive silicone elastomer material which is to be used in particular for the production of composite shaped articles or for casting electric/electronic parts, namely:

- a) good flowability and shelf-life,
- b) high crosslinking rate at relatively low temperatures,
- c) high adhesive strength on organic plastics, metals and glasses,
- d) easy demoldability from vulcanization molds,
- e) toxicological safety,
- f) high level of performance characteristics, especially (transparency, noncorrosiveness, and good mechanical property profile.

DISCLOSURE OF INVENTION

It was therefore an object of the present invention to provide an addition-crosslinking silicone elastomer material which has good self-adhesion to organic plastics, metals and glasses, does not have the above disadvantages, and meets the
5 desired requirements as previously discussed.

BEST MODE FOR CARRYING OUT THE INVENTION

The invention relates to self-adhesive addition-crosslinking silicone compositions which comprise

(A) diorganopolysiloxane(s) of the general formula (1)



in which

R^1 is a hydroxyl radical or a monovalent, optionally halogen-substituted hydrocarbon radical optionally containing O, N, S or P atoms, having 1 to 20 carbon atoms, and being free of aliphatically unsaturated groups,

15 R^2 is a monovalent, aliphatically unsaturated, optionally halogen-substituted hydrocarbon radical optionally containing O, N, S or P atoms and having 2 to 10 carbon atoms,

b has a value from 0.003 to 2,

20 with the proviso that $1.5 < (a+b) < 3.0$, that on average at least two aliphatically unsaturated radicals R^2 are present per molecule and that the viscosity of the diorganopolysiloxane(s) (A), determined at 25°C, is 1 mPa·s to 40,000 Pa·s,

(B) organohydrogenpolysiloxane(s) of the general formula (2)



in which

25 R^3 is a monovalent aliphatically saturated hydrocarbon radical having 1 to 20 carbon atoms,

- R⁴ is (a) an optionally halogen-substituted monovalent hydrocarbon radical having 6 to 15 carbon atoms which contains at least one aromatic C₆-ring, and/or (b) a halogen-substituted, saturated monovalent hydrocarbon radical optionally containing O, N, S or P atoms and having 2 to 20 carbon atoms,
- 5 R⁵ is a bivalent, optionally halogen-substituted hydrocarbon radical Si-bonded at both ends, optionally containing O, N, S or P atoms and having 6 to 20 carbon atoms, and c, d, e and f denote positive numbers, with the proviso that the organohydrogenpolysiloxane (B) contains on average 3 to less than 20 SiH groups per molecule, that the relationship: $0.05 < 100 (d+e)/(c+d+e+f) < 12$
- 10 is fulfilled, and that the viscosity of the organohydrogenpolysiloxane (B), determined at 25°C, is 1 mPa·s to 100 Pa·s,
- (C) organosilicon compounds having epoxy groups and hydrolyzable groups, of the general formula (3)



- 15 and/or their partial hydrolysis products, in which
- R⁷ is a hydrogen radical, a hydroxyl radical or an optionally halogen- or cyano-substituted, saturated monovalent hydrocarbon radical optionally containing O, N, S or P atoms and having 1 to 20 carbon atoms,
- R⁸ is an optionally halogen-substituted monovalent hydrocarbon radical containing at least one epoxy group, optionally containing O, N, S or P atoms and having 20 2 to 20 carbon atoms,
- R⁹ is a hydrolyzable, monovalent optionally halogen-substituted hydrocarbon radical bonded to Si via an Si-O-C-, Si-O-N- or Si-N- link, optionally containing O, N, S or P atoms and having 1 to 20 carbon atoms,
- 25 with the proviso that $4 > g \geq 0$, $4 > h > 0$, $4 > i > 0$, $4 \geq (h+i) > 0$ and $4 \geq (g+h+i)$, and
- (D) a hydrosilylation catalyst.

Organohydrogenpolysiloxane (B) acts as an adhesion promoter and simultaneously as a crosslinking agent.

In particular, the present composition is distinguished by the fact that

- a) the crosslinking rate is scarcely reduced,
- b) the transparency of the crosslinked silicone elastomers is not impaired,
- c) there is no need to accept any disadvantageous changes in the mechanical elastomer properties,
- d) the adhesion-promoting component (B) simultaneously acts as a crosslinking agent (no additional SiH crosslinking agent required),
- e) strong self-adhesion can be achieved even on metals without hindering the demoldability from metal vulcanization molds (it was found that the adhesion to metal shortly after crosslinking permits demolding of the silicone elastomer part; if, however, the silicone elastomer/metal

composite is stored, the silicone elastomer grows strongly and permanently onto the metal surface within a short time),
f) the flowability of the uncrosslinked material is scarcely impaired.

Although the adhesion-promoting component (B) of the present invention
5 also has reduced compatibility with the other components of the material, which is evident from turbidity on mixing in, this turbidity disappears completely as soon as the material is heated for the purpose of crosslinking; this indicates a homogeneous distribution of the molecules of the crosslinking agent in the material at the time of crosslinking. If, on the other hand, the adhesion-promoting SiH-containing
10 component (B) contains phenyl groups 12 mol% or more of the radicals, turbidity also persists at customary crosslinking temperatures and indicates inhomogeneous network formation, which can also be demonstrated on the basis of the optical properties, crosslinking characteristics, and on the basis of the mechanical properties.

The components (A), (B) and (C) may each constitute a single compound
15 or a mixture of different compounds.

Examples of the radicals R^1 are alkyl radicals such as the methyl, ethyl, propyl, isopropyl, tert-butyl, n-pentyl, isopentyl, neopentyl, tert-pentyl, n-octyl, 2-ethylhexyl, 2,2,4-trimethylpentyl, n-nonyl, and octadecyl radicals; cycloalkyl radicals such as the cyclopentyl, cyclohexyl, cycloheptyl, norbornyl, adamantylethyl
20 and bornyl radicals; aryl or aralkyl radicals such as the phenyl, ethylphenyl, tolyl, xylyl, mesityl, and naphthyl radicals; aralkyl radicals such as the benzyl, 2-phenylpropyl and phenylethyl radicals; and those derivatives of the above radicals which are halogenated and functionalized with organic groups, such as the 3,3,3-trifluoropropyl, 3-iodopropyl, 3-isocyanatopropyl, aminopropyl,
25 methacryloyloxymethyl, and cyanoethyl radicals. Preferred radicals R^1 contain 1 to 10 carbon atoms and optionally contain halogen substituents. Particularly preferred radicals R^1 are the methyl, phenyl and 3,3,3-trifluoropropyl radicals, in particular the methyl radical.

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The radicals R^2 are obtainable by a hydrosilylation reaction. Examples of these are alkenyl and alkynyl radicals such as the vinyl, allyl, isopropenyl, 3-butenyl, 2,4-pentadienyl, butadienyl, 5-hexenyl, undecenyl, ethynyl, propynyl and hexynyl radicals; cycloalkenyl radicals such as the cyclopentenyl, cyclohexenyl, 3-cyclohexenylethyl, 5-bicycloheptenyl, norbornenyl, 4-cyclooctenyl, and cyclooctadienyl radicals; alkenylaryl radicals such as the styryl and styrylethyl radicals; and those derivatives of the above radicals which are halogenated and/or contain heteroatoms, such as the 2-bromovinyl, 3-bromo-1-propynyl, 1-chloro-2-methylallyl, 2-(chloromethyl)allyl, styryloxy, allyloxypropyl, 1-methoxyvinyl, cyclopentenyl, 3-cyclohexenyl, acryloyl, acryloyloxy, methacryloyl, and methacryloyloxy radicals. Preferred radicals R^2 are the vinyl, allyl and 5-hexenyl radicals, in particular the vinyl radical.

In the case of the diorganopolysiloxanes (A) of the general formula (1), the viscosity determined at 25°C is preferably 100 mPa·s to 30,000 Pa·s. More preferably, the viscosity range is from 1 to 30,000 Pa·s. Depending on the type of addition-crosslinking material, different viscosity ranges may be preferred. Viscosities from 100 to 10,000 mPa·s are particularly preferred for the materials known as RTV-2 (room temperature vulcanizing) compositions, from 1 to 100 Pa·s for LSR (liquid silicone rubber) compositions, and from 2000 to 40,000 Pa·s for HTV (high temperature vulcanizing) compositions.

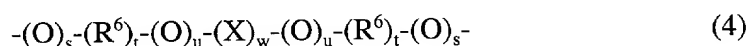
Examples of R^3 are alkyl radicals such as the methyl, ethyl, propyl, isopropyl, tert-butyl, n-octyl, 2-ethylhexyl and octadecyl radicals; and cycloalkyl radicals such as the cyclopentyl, cyclohexyl, norbornyl, and bornyl radicals. Preferred radicals R^3 are hydrocarbon radicals having 1 to 10 carbon atoms. A particularly preferred radical R^3 is the methyl radical.

Examples of R^4 (a) are the phenyl, tolyl, xylyl, biphenyl, anthryl, indenyl, phenanthryl, naphthyl, benzyl, phenylethyl and phenylpropyl radicals, and those derivatives of the above radicals which are halogenated and functionalized with organic groups, such as the o-, m- or p-chlorophenyl, pentafluorophenyl, bromotolyl, trifluorotolyl, phenoxy, benzyloxy, benzyloxyethyl, benzoyl, benzoyl-

oxy, p-tert-butylphenoxypropyl, 4-nitrophenyl, quinolinyl, and pentafluorobenzoyloxy radicals.

Examples of hydrocarbon radicals R^4 (b) having 2 to 20 carbon atoms are the 3-chloropropyl, 3-bromopropyl, 3,3,3-trifluoropropyl, 2-fluoroethyl, 1,1-dihydroperfluorododecyl, and 2-cyanoethyl radicals. Particularly preferred radicals R^4 are the phenyl radical and the 3,3,3-trifluoropropyl radical.

Preferred radicals R^5 correspond to the general formula (4)



in which

s, t, u and w, independently of one another, denote the values 0, 1 or 2,

R^6 may be identical or different and denotes a bivalent, optionally halogen-substituted hydrocarbon radical which optionally contains O, N, S or P atoms, is free of aliphatically unsaturated aliphatic groups and contains 1 to 10 carbon atoms, such as $-\text{CH}_2-$, $-\text{CH}_2-\text{CH}_2-$, $-\text{CH}_2-\text{CH}_2-\text{CH}_2-$, $-\text{CF}_2-$, $-\text{CH}_2-\text{CF}_2-$, $-\text{CH}_2-\text{CH}(\text{CH}_3)-$, $-\text{C}(\text{CH}_3)_2-$, $-\text{CH}_2-\text{C}(\text{CH}_3)_2-$, $-\text{C}(\text{CH}_3)_2-\text{CH}_2-$, $-\text{CH}_2-\text{CH}_2-\text{O}-$ or $-\text{CF}_2-\text{CF}_2-\text{O}-$,

$-(X)-$ denotes a bivalent radical which may be $-\text{Ph}-$, $-\text{Ph}-\text{O}-\text{Ph}-$, $-\text{Ph}-\text{S}-\text{Ph}-$, $-\text{Ph}-\text{SO}_2-\text{Ph}-$, $-\text{Ph}-\text{C}(\text{CH}_3)_2-\text{Ph}-$, $-\text{Ph}-\text{C}(\text{CF}_3)_2-\text{Ph}-$, $-\text{Ph}-\text{C}(\text{O})-\text{Ph}-$, cyclohexylene or norbornylene, $-\text{Ph}-$ designating a phenylene group. A particularly preferred radical R^5 is the phenylene radical.

The organohydrogenpolysiloxane (B) preferably contains 5 to 18 SiH groups per molecule. The viscosity of the component (B), measured at 25°C, is preferably 2 mPa·s to 1 Pa·s. Owing to the labile nature of the SiH group, the component (B) may have a low content, typically <100 ppm by weight, of Si-bonded OH groups, due to the method of preparation.

Examples of hydrocarbon radicals R^7 are alkyl radicals such as the methyl, ethyl, n-propyl, isopropyl, tert-butyl, n-pentyl, isopentyl, neopentyl, tert-pentyl,

25 Examples of the radicals R^8 are the epoxyethyl, 2,3-epoxypropyl, 3,4-epoxybutyl, 5,6-epoxyhexyl, 9,10-epoxydecyl, glycidyloxy, 3-glycidyloxypropyl, glycidyloxyisobutyl, 2-methylglycidyloxypropyl, 3-phenylglycidyloxypropyl, glycidyloxyphenylnonyl, glycidyloxybenzylethyl, 3,4-epoxycyclohexyl, 2-(3,4-epoxycyclohexyl)ethyl, 3-(3,4-epoxycyclohexyl)propyl, 1,4-epoxycyclohexyl, and
30 2-(1,4-epoxycyclohexyl)ethyl radicals. Preferred radicals R^8 are the 3,4-epoxycyclohexyl, 3-(3,4-epoxycyclohexyl)propyl and glycidyloxypropyl radicals. R^8

preferably has 2 to 10 carbon atoms. A particularly preferred radical R^8 is the glycidylpropyl radical.

R^9 denotes a hydrolyzable, monovalent, optionally halogen-substituted hydrocarbon radical bonded to Si via an Si-O-C-, Si-O-N- or Si-N-link, optionally containing O, N, S or P atoms and having 1 to 20 carbon atoms.

Examples of the radicals R^9 are

- a) alkoxy, alkenoxy or aryloxy groups of the general formula $-OR^{10}$, such as the methoxy, ethoxy, n-propoxy, isopropoxy, n-butoxy, sec-butoxy, tert-butoxy, 2-ethylbutoxy, 2-ethylhexyloxy, vinyloxy, allyloxy, isopropenyloxy, cyclobutenyloxy, cyclohexenyloxy, 1,3-butadienyloxy, propargyloxy, phenoxy, benzyloxy, and m- and p-vinylbenzyloxy radicals;
- b) acyloxy groups of the general formula $-OCOR^{10}$, such as the formyloxy, acetoxy, 2-ethylhexanoyloxy, acryloyloxy, methacryloyloxy, benzoyloxy, and norbornyl-acetoxy radicals;
- c) amino groups of the general formula $-NH_2$, $-NHR^{10}$, and $-NR^{10}_2$, such as the dimethylamino, diisopropylamino, allylamino, n-butylamino, sec-butylamino, and cyclohexylamino radical;
- d) oxime groups of the general formula $-ON=CH_2$, $-ON=CHR^{10}$, and $-ON=CR^{10}_2$, such as the methyl ethyl ketoxime, methyl isobutyl ketoxime, methyl n-amyl ketoxime, and dimethyl ketoxime radicals;
- e) amido groups of the general formula $-NH-C(=O)-R^{10}$ or $-NR^{10}-C(=O)-R^{10}$, such as the N-methylbenzamido and N-methylacetamido radicals;
- f) aminoxy groups of the general formula $-ONH_2$, $-ONHR^{10}$ or $-ONR^{10}_2$, such as the hydroxylamino radical; and
- g) those derivatives of the abovementioned radicals which are halogen-substituted or contain heteroatoms or have an otherwise complex composition, such as p-aminophenoxy, 2-methoxyethoxy, 1-methoxy-2-propoxy, 1-methoxy-isopropenyloxy, methoxyethoxyethoxy, 1-methoxy-2-methylpropenyloxy, acryloyloxymethoxy, methacryloyloxy(polyethyleneoxy), furyloxy, and N-vinyl-formamido radicals, and $-O-Ph-C(=O)-Ph$, $-O-C(CF_3)=CH-C(=O)-CF_3$, $-O-C(CH_3)=CH-C(=O)-CH_3$, $-O-C(CH_3)_2-CH=CH_2$, $-NH-C(=O)-CH_3$,

-O-C(=O)-CH₂Br, -O-C(=O)-CF₃, -O-C(=O)-C≡CH or -O-CH₂-C(=O)-O-Si(CH₃)₃,

where the radicals R¹⁰ represent monovalent aliphatic or aromatic, saturated or unsaturated, optionally halogen-substituted hydrocarbon radicals having 1 to 10 carbon atoms. Preferred radicals R⁹ are the alkoxy radicals such as the methoxy, ethoxy, propoxy, and butoxy radicals. The particularly preferred radical R⁹ is the methoxy radical. A particularly suitable organosilicon compound (C) is glycidyloxypropyltrimethoxysilane (Glymo).

The radicals R¹ to R¹⁰ in all above formulae may be identical or different. Preferred heteroatoms are N, O and S. Preferred halogen substituents are F, Cl and Br.

Preferably 0.1 to 50 parts by weight, in particular 0.5 to 10 parts by weight, of organohydrogenpolysiloxane (B) and 0.1 to 10 parts by weight, in particular 0.5 to 5 parts by weight, of organosilicon compound (C) are used per 100 parts by weight of diorganopolysiloxane (A).

Hydrosilylation catalyst (D) serves as a catalyst for the hydrosilylation addition reaction between the aliphatically unsaturated hydrocarbon radicals R² of the diorganopolysiloxanes (A) and the silicon-bonded hydrogen atoms of the organohydrogenpolysiloxanes (B). Numerous suitable hydrosilylation catalysts are described in the literature. In principle, all hydrosilylation catalysts corresponding to the prior art and used in addition-crosslinking silicone rubber materials can be used.

Metals and their compounds, such as platinum, rhodium, palladium, ruthenium and iridium, preferably platinum, can be used as hydrosilylation catalysts (D). The metals can optionally be fixed on finely divided support materials, such as active carbon, metal oxides, such as alumina, or silica.

Platinum and platinum compounds are preferably used. Particularly preferred platinum compounds are those which are soluble in polyorganosiloxanes.

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The soluble platinum compounds used may be, for example, the platinum-olefin complexes of the formulae $(\text{PtCl}_2 \cdot \text{olefin})_2$ and $\text{H}(\text{PtCl}_3 \cdot \text{olefin})$, alkenes having 2 to 8 carbon atoms such as ethylene, propylene, isomers of butene and of octene, and cycloalkanes having 5 to 7 carbon atoms such as cyclopentene, cyclohexene and cycloheptene preferably being used. Further soluble platinum catalysts are the platinum-cyclopropane complexes of the formula $(\text{PtCl}_2 \text{C}_3\text{H}_6)_2$, the reaction products of hexachloroplatinic acid with alcohols, ethers and aldehydes and mixtures thereof or the reaction product of hexachloroplatinic acid with methylvinylcyclotetrasiloxane in the presence of sodium bicarbonate in ethanolic solution. Platinum catalysts with phosphorous, sulfur and amine ligands may also be used, e.g. $(\text{Ph}_3\text{P})_2\text{PtCl}_2$. Complexes of platinum with vinylsiloxanes, such as sym-divinyltetramethyldisiloxane, are particularly preferred.

The amount of hydrosilylation catalyst (D) used depends on the desired crosslinking rate and economic points of view. Preferably, 1×10^{-5} to 5×10^{-2} part by weight, in particular 1×10^{-3} to 1×10^{-2} part by weight of platinum catalysts, calculated as platinum metal, is used per 100 parts by weight of diorganopolysiloxanes (A).

The self-adhesive addition-crosslinking silicone compositions can optionally contain conventional further components (E), such as fillers, inhibitors, stabilizers, pigments and catalysts.

In order to achieve a sufficiently high mechanical strength of the crosslinked silicone rubber, it is preferable to incorporate actively reinforcing fillers as component (F) into the addition-crosslinking silicone compositions. The actively reinforcing fillers (F) preferably used are, in particular, precipitated and pyrogenic silicas, and mixtures thereof. The specific surface area of these actively reinforcing fillers should be at least $50 \text{ m}^2/\text{g}$ or preferably in the range from 100 to $400 \text{ m}^2/\text{g}$ according to the determination by the BET method. Such actively reinforcing fillers are very well known materials in the area of silicone rubbers.

5 The compounding of the self-adhesive addition-crosslinking silicone compositions is effected by mixing the abovementioned ingredients in any desired sequence. The crosslinking of the self-adhesive addition-crosslinking silicone compositions is preferably effected by heating, preferably at 30°C to 250°C, preferably at at least 50°C, in particular at at least 100°C, and preferably at not more than 200°C, more preferably at not more than 180°C.

The invention also relates to the addition-crosslinked silicone elastomers prepared from the crosslinkable compositions.

10 The silicone compositions can be bonded to a substrate by applying the silicone compositions to the substrate and then crosslinking them, preferably by heating to 30 to 250°C, to give a composite material. The self-adhesive addition-crosslinking silicone composition can advantageously be used in particular where good adhesive strength between the addition-crosslinked silicone elastomer and a substrate, preferably an organic polymer, metal or glass substrate, is desired. The
15 substrate may be present as a shaped article, film or coating. The self-adhesive addition-crosslinking silicone compositions are suitable for the production of composite materials by coating, adhesive bonding or casting, and for the production of shaped articles.

20 The self-adhesive addition-crosslinking silicone compositions are particularly suitable for casting and for adhesively bonding electrical and electronic parts and for the production of composite shaped articles. Composite shaped articles are understood here as meaning a uniform shaped article comprising a composite material which is composed of a silicone elastomer part produced from the silicone compositions and at least one substrate, so that there is a strong, permanent bond
25 between the two parts. Such a composite shaped article is preferably produced by processing an organic polymer or polymer precursor to give a shaped article, followed by bringing the silicone composition into contact with this shaped article and crosslinking. This can be effected, for example, by injection molding, by means of extrusion, and by the so-called press-molding method. Composite materials and
30 in particular composite shaped articles can be used in a very wide range of

applications, for example in the electronics, household appliance, consumables, construction, and automotive industries, in medical technology, in the production of sport and leisure articles, etc.

5 In the following examples, unless stated otherwise in each case, all pressures are 0.10 MPa (abs.), and all temperatures are 20°C.

Examples:

Substrates

10 The adhesion of the addition-crosslinked silicone elastomers according to the invention and comparative elastomers not according to the invention was tested on the following substrates:

- a) polybutylene terephthalate (PBT): Ultradur® B4300G6 (BASF AG; 30 % GF)
- b) polyamide 6: Durethan® BKV30 (Bayer AG; 30 % GF)
- c) polycarbonate (PC): Lexan® (General Electric)
- d) aluminum (industrial quality; not primed)
- 15 e) steel: VA stainless steel (industrial quality).

Characterization of adhesion

(Examples 1, 2, and 3 and Comparative Examples C1 and C2)

20 To characterize adhesive strength, 100.0 g of an addition-crosslinking liquid silicone elastomer material which is free of crosslinking agent, and contains 71.6% by weight of diorganopolysiloxane having a viscosity of 20,000 mPa·s and 28.4% by weight of a reinforcing filler and is available from Wacker-Chemie GmbH under the name ELASTOSIL® LR 3003/40 A, are thoroughly mixed with a specified amount of the adhesion-promoting additive which is prepared according to the following examples and which simultaneously serves as a crosslinking agent, and
25 then degassed in vacuo at room temperature. This material is then immediately applied as an about 1.2 mm thick layer, by means of a doctor blade, to the respective substrate surface which has been cleaned beforehand with ethanol. A primed, hollow

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aluminum article having a contact area of 100 mm² is then placed on the still uncrosslinked layer of the liquid silicone elastomer material. The priming of the hollow aluminum article guarantees bonding of the crosslinked silicone elastomer to be strong in each case, so that under tensile stress, either a loss of cohesion within the silicone elastomer occurs or a loss of adhesion between silicone elastomer and substrate occurs. By means of a spacer and by slight pressure on the hollow aluminum article, a defined layer thickness of 1 mm is set. The laminate produced in this manner and consisting of substrate, 1 mm thick liquid silicone elastomer material and hollow aluminum article, is then vulcanized at a temperature of 130°C for 30 min in a forced-circulation drying oven, complete crosslinking of the liquid silicone elastomer material taking place, followed by cooling to room temperature. By means of a tensile tester, the maximum force which is required to separate the hollow aluminum article completely from the substrate, i.e. to break the adhesive bond, is then measured. The maximum value of this force, based on the contact area of 100 mm², is stated as the breaking tension in N/mm². Ten laminates are measured per example, the breaking tension is determined as a mean value and the proportion of cohesive failure is determined as a percentage. A cohesive failure of 0% means that the silicone elastomer was detached from the substrate surface completely and without leaving a residue. A cohesive failure of 100% means that the delamination took place exclusively by crack growth within the silicone elastomer.

Example 1

5.0 kg of a methylhydrogenpolysiloxane, $\text{Me}_3\text{Si}(-\text{O}-\text{SiH}(\text{Me}))_n-\text{O}-\text{SiMe}_3$, which has terminal trimethylsilyloxy groups and, according to ²⁹Si-NMR, has a number average chain length of $n=53$, and 5.0 kg of a siloxane copolymer having terminal trimethylsilyloxy groups and consisting of dimethylsilyloxy and methylphenylsilyloxy units in a molar ratio of 15.2:21.8 and having a viscosity, measured at 25°C, of 100 mPa·s, are initially introduced into a stirrer. 1.8 g of phosphonitrilic chloride, prepared according to US-A-4,203,913, are added at room temperature with continuous stirring. After application of a vacuum of <20 mbar, stirring is carried out for 5 min and the vacuum is then broken with N₂. Stirring is then carried out for 2 h at a temperature of 100°C. 14 g of hexamethyldisilazane are then added and stirring is carried out for a further 15 min. Finally, a vacuum of <10

mbar is applied, the mixture is freed from volatile components with continuous stirring for 1 h at 100°C, the vacuum is broken with N₂, cooling to room temperature is effected, and the reaction product is filtered. A polyorganosiloxane which has terminal trimethylsilyloxy groups, is composed of -O-Si(Me)₂-, -O-SiH(Me)- and -O-Si(Me)Ph- units in a molar ratio of 15.2:63.0:21.8 and has a viscosity of 32.1 mm²/s at 25°C is obtained. The product contains on average 17 Si-H groups per molecule.

The preparation of the adhesion-promoting additive, which is used in an amount of 1.3 g per 100 g of silicone elastomer material to be crosslinked, is effected by mixing 10 parts by weight of glycidylxypropyltrimethoxysilane with 16 parts by weight of the reaction product described above. The results of the adhesion measurements are shown in Table 1.

Comparative Example C1

Instead of the phenyl-containing crosslinking agent described in Example 1, a polyorganosiloxane which has terminal trimethylsilyloxy groups, contains only -O-Si(Me)₂- and -O-SiH(Me)- units in a molar ratio of 29:71 and has a viscosity of 30 mm²/s at 25°C is used; this crosslinking agent contains on average 33 Si-H groups per molecule. The preparation of the adhesion-promoting additive, which is used in an amount of 1.05 g per 100 g of silicone elastomer material to be crosslinked, is effected by mixing 10 parts by weight of glycidylxypropyltrimethoxysilane with 11 parts by weight of the phenyl-free crosslinking agent described above. This mixing ratio and the amount of adhesion-promoting additive used ensure that the addition-crosslinking silicone elastomer materials prepared according to Example 1 and Example 2 have the same content of Si-bonded hydrogen. The results of the adhesion measurements are shown in Table 1.

Example 2

0.05 g of phosphonitrilic chloride, prepared according to US-A-4,203,913, is added at room temperature with continuous stirring, to 150 g of a methylhydrogen-polysiloxane, Me₃Si-(-O-SiH(Me))_n-O-SiMe₃, which has terminal trimethylsilyloxy groups and, according to ²⁹Si-NMR, has a number average chain length of n=53, and 150 g of a siloxane copolymer having terminal trimethylsiloxy groups, consisting

of dimethylsilyloxy and diphenylsilyloxy units, having a phenyl content of 19.2% by weight and having a viscosity of 85 mPa·s, measured at 25°C. After application of a vacuum of <20 mbar, stirring is carried out for 5 min and the vacuum is then broken with N₂. Stirring is then carried out for 2 h at a temperature of 100°C. 0.5 g of hexamethyldisilazane is added and stirring is carried out for a further 15 min. Finally, a vacuum of <10 mbar is applied, the mixture is freed from volatile components with continuous stirring for 1 h at 100°C, the vacuum is broken with N₂, cooling to room temperature is effected and the reaction product is filtered. A polyorganosiloxane which has terminal trimethylsilyloxy groups, is composed of -O-Si(Me)₂-, -O-SiH(Me)- and -O-SiPh₂- units in a molar ratio of 26.5:70.0:3.5 and has a viscosity of 455 mm²/s at 25°C is obtained.

The preparation of the adhesion-promoting additive, which is used in an amount of 1.3 g per 100 g of silicone elastomer material to be crosslinked, is effected by mixing 10 parts by weight of glycidyloxypropyltrimethoxysilane with 16 parts by weight of the reaction product described above.

The results of the adhesion measurements are shown in Table 1.

Example 3

150 g of a methylhydrogenpolysiloxane, Me₃Si-(O-SiH(Me))_n-O-SiMe₃, which has terminal trimethylsilyloxy groups and, according to ²⁹Si-NMR, has a number average chain length of n=53, 26 g of a polydimethylsiloxane which has terminal trimethylsilyloxy groups and has a viscosity of 100 mPa·s at 25°C, and 112 g of 1,3,5-trimethyl-1,3,5-tris(3,3,3-trifluoropropyl)cyclotrisiloxane are initially introduced at room temperature and mixed, with continuous stirring, with 0.05 g of phosphonitrilic chloride, prepared by US-A-4,203,913. After application of a vacuum of <20 mbar, stirring is carried out for 5 min and the vacuum is then broken with N₂. Stirring is then carried out for 2 h at a temperature of 100°C. In order to deactivate the catalyst (phosphonitrilic chloride), gaseous ammonia is passed in for 15 min. Thereafter, a vacuum of <10 mbar is applied, the mixture is freed from volatile components with continuous stirring for 1 h at 100°C, the vacuum is broken with N₂, cooling to room temperature is effected, and the reaction product is filtered. A polyorganosiloxane which has terminal trimethylsilyloxy groups,

composed of $\text{-O-Si(Me)}_2\text{-}$, -O-SiH(Me)- and $\text{-O-Si(Me)(CH}_2\text{CH}_2\text{CF}_3\text{)-}$ units in a molar ratio of 10.1:69.9:20.0 and which has a viscosity of $52.3 \text{ mm}^2/\text{s}$ at 25°C is obtained.

5 The preparation of the adhesion-promoting additive, which is used in an amount of 1.3 g per 100 g of silicone elastomer material to be crosslinked, is effected by mixing 10 parts by weight of glycidyoxypropyltrimethoxysilane with 16 parts by weight of the reaction product described above. The results of the adhesion measurements are shown in Table 1.

Comparative Example C2

10 The composition corresponds to the silicone elastomer material described in Example 1, except that no glycidyoxypropyltrimethoxysilane was added. The results of the adhesion measurements are shown in Table 1.

Table 1: (breaking tension in $[\text{N}/\text{mm}^2]$; proportion of cohesive failure in $[\%]$)

Ex.	PBT		Polyamide 6		PC		Aluminum		Steel	
	$[\text{N}/\text{mm}^2]$	$[\%]$	$[\text{N}/\text{mm}^2]$	$[\%]$	$[\text{N}/\text{mm}^2]$	$[\%]$	$[\text{N}/\text{mm}^2]$	$[\%]$	$[\text{N}/\text{mm}^2]$	$[\%]$
1	2.72	100	3.05	50	2.38	0	2.89	100	2.65	100
C1	1.60	0	1.41	0	1.51	0	1.90	0	1.98	0
2	2.56	100	3.12	30	2.33	0	3.23	100	2.55	90
3	2.22	80	2.78	40	2.08	30	2.78	100	2.25	90
C2	1.20	10	1.53	20	1.87	0	2.20	20	2.13	20

20 The values shown in Table 1 demonstrate the high adhesive strength between the addition-crosslinked silicone elastomer according to the invention (Examples 1, 2, and 3) and various organic plastics and metals. As is evident from Comparative Example C1, high adhesive strength is not achieved by the presence of component (C) alone (e.g. glycidyoxypropyltrimethoxysilane) but only in conjunction with
25 component (B) according to the invention. As is evident from Comparative Example C2, a high adhesive strength is not achieved by the presence of the component (B)

alone but only in conjunction with component (C) according to the invention. This is evidence of a synergistic effect of the adhesion-promoting components (B) and (C) of the silicone elastomer material according to the invention.

Characterization of the adhesion

5 (Examples 4, 5, 6 and Comparative Examples C3 and C4)

10 A substrate strip having dimensions $60 \times 20 \times 2$ mm is introduced into a stainless steel compression mold and the mold is filled with the addition-crosslinking silicone elastomer material to be tested. A press cure is carried out for 5 min at a temperature of 170°C and a pressure of 30 tonnes. After removal of the laminate, the rigid substrate element is firmly clamped and the maximum parting force required to detach the adhering silicone elastomer strip in a peel test is determined. The parting force is determined according to DIN 53531 and is stated in N/mm. Ten laminates are measured per example, the parting force is determined as a mean value, and the proportion of cohesive failure is determined as a percentage. Cohesive failure of 0% means that the silicone elastomer was detached completely and without residue from the substrate surface. Cohesive failure of 100% means that the delamination took place exclusively by crack growth within the silicone elastomer.

Example 4

20 The composition corresponds to the addition-crosslinking silicone elastomer material according to the invention described in Example 1. The results of the measurements of the parting force are shown in Table 2.

Example C3

25 The composition corresponds to the addition-crosslinking silicone elastomer material, not according to the invention, described in Comparative Example C1. The results of the measurements of the parting force are shown in Table 2.

Example 5

The composition corresponds to the addition-crosslinking silicone elastomer material according to the invention described in Example 2. The results of the measurements of the parting force are shown in Table 2.

5 **Example 6**

The composition corresponds to the addition-crosslinking silicone elastomer material according to the invention, described in Example 3. The results of the measurements of the parting force are shown in Table 2.

Comparative Example C4

10 The composition corresponds to the addition-crosslinking silicone elastomer material not according to the invention, described in Comparative Example C2. The results of the measurements of the parting force are shown in Table 2.

Table 2: (parting force in [N/mm]; proportion of cohesive failure in [%])

Ex.	PBT		Polyamide 6		Aluminum		Steel	
	[N/mm]	[%]	[N/mm]	[%]	[N/mm]	[%]	[N/mm]	[%]
4	14.8	100	9.3	30	14.6	100	16.4	100
C3	7.1	20	4.9	0	7.2	10	12.8	80
5	12.3	100	11.0	30	13.3	100	15.5	100
6	11.8	80	7.7	40	10.0	90	14.4	100
C4	6.2	0	4.4	0	8.9	30	10.5	40

20 The values shown in Table 2 demonstrate the high adhesive strength of laminates consisting of the addition-crosslinked silicone elastomer according to the invention (Examples 4, 5 and 6) and organic plastics and metals, for the case of press cure. As is evident from Comparative Example C3, high adhesive strength is not achieved by the presence of the component (C) alone (e.g. glycidylxypropyltrimethoxysilane) but
25 only in conjunction with component (B) according to the invention. As is evident from Comparative Example C4, furthermore, high adhesive strength is achieved not

by the presence of the component (B) alone but only in conjunction with a component (C) according to the invention. This is evidence of a synergistic effect of the adhesion-promoting components (B) and (C) of the silicone elastomer material according to the invention.

5 Vulcanization characteristics, optical and mechanical properties and compression set of addition-crosslinking silicone materials according to the invention and not according to the invention (Example 7 and Comparative Examples C5 and C6)

10 To assess the property profile of the self-adhesive silicone elastomer materials according to the invention and not according to the invention, 200.0 g of an addition-crosslinking liquid silicone elastomer material which is free of crosslinking agent, containing 71.6% by weight of diorganopolysiloxane having a viscosity of 20,000 mPa·s and 28.4% by weight of reinforcing filler, obtainable from Wacker-Chemie GmbH under the name ELASTOSIL® LR 3003/40 A, were thoroughly mixed with an amount, described in the following Examples C5, 7, and 15 C6, of adhesion-promoting additive which simultaneously serves as crosslinking agent, degassed under a vacuum of 10 mbar, filled into a compression mold and crosslinked at a temperature of 175°C for 10 min to give a 2 mm thick silicone elastomer film. The film was characterized with respect to transparency and mechanical properties (Shore A according to DIN 53505; elongation at break and 20 tensile strength according to DIN 53504; resilience according to DIN 53512; tear propagation strength according to ASTM D 624 B; compression set according to DIN 35517, 22 h / 175°C). Furthermore, the crosslinking characteristics of the freshly prepared uncrosslinked silicone material was determined by vulcametry (initiation temperature; crosslinking rate; maximum torque).

25 Comparative Example C5

2.1 g of the adhesion-promoting additive described in Comparative Example C1 are used according to the process described above. The results are shown in Tables 3 and 4.

Example 7

2.6 g of the adhesion-promoting additive described in Example 1 are used according to the process described above. The results are shown in Tables 3 and 4.

Comparative Example C6

- 5 0.18 g of phosphonitrilic chloride, prepared according to US-A-4,203,913, is added, at room temperature, with continuous stirring, to 250 g of a methylhydrogenpoly-siloxane, $\text{Me}_3\text{Si}(-\text{O}-\text{SiH}(\text{Me}))_n-\text{O}-\text{SiMe}_3$, which has terminal trimethylsilyloxy groups and, according to ^{29}Si -NMR, has a number average chain length of $n=53$, and 750
- 10 g of a siloxane copolymer having terminal trimethylsilyloxy groups, consisting of dimethylsilyloxy and methylphenylsilyloxy units and having a viscosity of 100 mPa·s, measured at 25°C. After application of a vacuum of <20 mbar, stirring is carried out for 5 min and the vacuum is then broken with N_2 . Stirring is then carried out for 2 h at a temperature of 100°C. 1.5 g of hexamethyldisilazane are added and stirring is carried out for a further 15 min. Finally, a vacuum of <10 mbar is applied, the
- 15 mixture is freed from volatile components with continuous stirring for 1 h at 100°C, the vacuum is broken with N_2 , cooling to room temperature is effected, and the reaction product is filtered. 933 g of a clear silicone oil having a viscosity of 175 mPa·s and a content of 0.39% by weight of Si-bonded hydrogen are obtained. According to the ^{29}Si -NMR spectrum, the reaction product is a siloxane copolymer
- 20 having terminal trimethylsilyloxy groups and consisting of $-\text{O}-\text{Si}(\text{Me})_2-$, $-\text{O}-\text{SiH}(\text{Me})-$ and $-\text{O}-\text{Si}(\text{Me})\text{Ph}-$ units in a molar ratio of 25.9:36.2:37.9. The preparation of the adhesion-promoting additive which is used in an amount of 4.0 g according to the process described above is carried out by mixing 10 parts by weight of glycidyoxypropyltrimethoxysilane with 30 parts by weight of the reaction product
- 25 described above. The results are shown in Tables 3 and 4.

Table 3: Transparency and crosslinking characteristics

5

Example	Transparency	Initiation temperature	Crosslinking rate	Maximum torque
		[°C]	[Nm/min]	[Nm]
C5	translucent	105	1.09	0.46
7	translucent	114	0.52	0.36
C6	milky turbid	103	0.13	0.19

10

The results of the measurements shown in Table 3 demonstrate the disadvantages with regard to transparency and crosslinking characteristics of comparative self-adhesive silicone elastomer materials not according to the invention, whose SiH-containing component is distinguished by higher contents of incompatible radicals (> 12 mol%). In addition to increasing turbidity of the crosslinked silicone elastomer, there is also increasing inhibition of the crosslinking reaction, necessitating a limitation of the content of incompatible radicals of the component (B) to < 12 mol%, as in the silicone materials according to the invention.

Table 4: Mechanical properties

15

Example	Shore A	Elongation at break	Tensile strength	Resilience	Tear propagation strength	Compression set
		[%]	[N/mm ²]	[%]	[N/mm]	[%]
C5	39	650	8.6	54	25.0	44
7	34	750	8.7	49	23.8	48
C6	26	950	8.5	36	33.7	75

20

The values of the mechanical properties of the vulcanized products, shown in Table 4, demonstrate the considerable impairment of the crosslinking which occurs in the case of a very high content of incompatible radicals in the SiH crosslinking agent,

which impairment is reflected in a low network density (low Shore A, greatly reduced resilience) and the persistence of numerous reactive groups, in particular SiH groups (high values of the compression set).

- 5 While embodiments of the invention have been illustrated and described, it is not intended that these embodiments illustrate and describe all possible forms of the invention. Rather, the words used in the specification are words of description rather than limitation, and it is understood that various changes may be made without departing from the spirit and scope of the invention.

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WHAT IS CLAIMED IS:

1 1. A self-adhesive addition-crosslinking silicone composition, comprising
2 (A) diorganopolysiloxane(s) of the general formula (1)



4 in which

5 R^1 is a hydroxyl radical or a monovalent, optionally halogen-substituted C_{1-20}
6 hydrocarbon radical optionally containing O, N, S or P atoms and free of
7 aliphatically unsaturated groups,

8 R^2 is a monovalent, aliphatically unsaturated, optionally halogen-substituted C_{2-10}
9 hydrocarbon radical optionally containing O, N, S or P atoms,

10 b has a value from 0.003 to 2,

11 with the proviso that $1.5 < (a+b) < 3.0$, that on average at least two aliphatically
12 unsaturated radicals R^2 are present per molecule, and that the viscosity of the
13 diorganopolysiloxane(s) (A), determined at 25°C, is 1 mPa·s to 40,000 Pa·s;

14 (B) organohydrogenpolysiloxane(s) of the general formula (2)



16 in which

17 R^3 is a monovalent aliphatically saturated C_{1-20} hydrocarbon radical,

18 R^4 is (a) an optionally halogen-substituted monovalent C_{6-15} hydrocarbon radical
19 which contains at least one aromatic C_6 -ring, or

20 (b) a halogen-substituted, saturated monovalent C_{2-20} hydrocarbon radical
21 optionally containing O, N, S or P atoms,

22 R^5 is a bivalent, optionally halogen-substituted C_{6-20} hydrocarbon radical Si-bonded
23 at both ends, optionally containing O, N, S or P atoms,

24 c, d, e and f denote positive numbers, with the proviso that the
25 organohydrogenpolysiloxane (B) contains on average 3 to less than 20 SiH groups per
26 molecule, that the relationship: $0.05 < 100 (d+e)/(c+d+e+f) < 12$ is fulfilled, and

27 that the viscosity of the organohydrogenpolysiloxane (B), determined at 25°C, is 1
28 mPa·s to 100 Pa·s;

29 (C) organosilicon compound(s) having epoxy groups and hydrolyzable groups of the
30 general formula (3)



32 and/or their partial hydrolysis products, in which

33 R^7 is a hydrogen radical, a hydroxyl radical or an optionally halogen- or cyano-
34 substituted, saturated monovalent C_{1-20} hydrocarbon radical optionally containing
35 O, N, S or P atoms,

36 R^8 is an optionally halogen-substituted monovalent C_{2-20} hydrocarbon radical
37 containing at least one epoxy group, optionally containing O, N, S or P atoms,

38 R^9 is a hydrolyzable, monovalent optionally halogen-substituted C_{1-20} hydrocarbon
39 radical bonded to Si via an Si-O-C-, Si-O-N- or Si-N- link, optionally containing
40 O, N, S or P,

41 with the proviso that $4 > g \geq 0$, $4 > h > 0$, $4 > i > 0$, $4 \geq (h+i) > 0$ and $4 \geq (g+h+i)$;
42 and

43 (D) a hydrosilylation catalyst.

1 2. The self-adhesive addition-crosslinking silicone composition of claim 1,
2 wherein the viscosity of the component (B) measured at 25°C, is 2 mPa·s to 1 Pa·s.

1 3. A process for the preparation of self-adhesive addition-crosslinked
2 silicone elastomers, in which the self-adhesive addition-crosslinking silicone
3 compositions of claim 1 is heated to 30°C to 250°C.

1 4. A process for the preparation of self-adhesive addition-crosslinked
2 silicone elastomers, in which the self-adhesive addition-crosslinking silicone
3 compositions of claim 2 is heated to 30°C to 250°C.

1 5. A self-adhesive addition-crosslinked silicone elastomer obtained by the
process of claim 3.

1 6. A process for the preparation of self-adhesive addition-crosslinked
2 silicone elastomers, in which the self-adhesive addition-crosslinking silicone
3 compositions of claim 4 is heated to 30°C to 250°C.

1 7. A process for bonding an addition-crosslinkable silicone composition
2 to a substrate, in which the self-adhesive addition-crosslinkable silicone compositions
3 of claim 1 is applied to the substrate and crosslinked by heating to 30°C to 250°C.

1 8. A process for bonding an addition-crosslinkable silicone composition
2 to a substrate, in which the self-adhesive addition-crosslinkable silicone compositions
3 of claim 2 is applied to the substrate and crosslinked by heating to 30°C to 250°C.

1 9. A composite material obtained by the process of claim 7.

1 10. A composite material obtained by the process of claim 8.

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ABSTRACT OF THE DISCLOSURE

The invention relates to self-adhesive addition-crosslinking silicone compositions which comprises

- (A) diorganopolysiloxanes, (B) organohydrogenpolysiloxanes containing less than 20
5 SiH bonded hydrogen atoms and less than 12% aryl groups, (C) organosilicon
compounds having epoxy groups and hydrolyzable groups and (D) a hydrosilylation
catalyst. The compositions are transparent to translucent, exhibit good compositional
compatibility, and cure to form strongly adhesive elastomers with good physical
properties.

DECLARATION FOR PATENT APPLICATION AND POWER OF ATTORNEY

Atty. Docket No. WAS 0394 PUS
First Named Inventor Philipp Müller et al.

As a below named inventor, I hereby declare that my residence, post office address and citizenship are as stated below next to my name.

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled:

SELF-ADHESIVE ADDITION-CROSSLINKING SILICONE COMPOSITIONS

the specification of which:

☒ [X] is attached hereto; or
☐ [] was filed on (MM/DD/YYYY) ____ as U.S. Application Number or PCT International Application Number ____, and was amended on (MM/DD/YYYY) ____ (if applicable).

I hereby state that I have reviewed and understand the contents of the above-identified specification, including the claims, as amended by any amendment specifically referred to above.

I acknowledge the duty to disclose information which is material to the examination of this application in accordance with Title 37, Code of Federal Regulations, §1.56.

I hereby claim foreign priority benefits under Title 35, United States Code, § 119(a)-(d) or § 365(b) of any foreign application(s) for patent or inventor's certificate, or § 365(a) of any PCT international application which designated at least one country other than the United States of America, listed below, and have also identified below, by checking the box, any foreign application for patent or inventor's certificate, or of any PCT international application having a filing date before that of the application on which priority is claimed.

<i>Prior Foreign Application Number(s)</i>	<i>Country</i>	<i>Foreign Priority Date (MM/DD/YYYY)</i>	<i>Priority Not Claimed</i>	<i>Certified Copy Attached? (Yes/No)</i>
199 59 412.0	Germany	December 9, 1999	Yes	Yes

I hereby claim the benefit under Title 35, United States Code, § 119(e) of any United States provisional application(s) listed below.

<i>Application Number(s)</i>	<i>Filing Date (MM/DD/YYYY)</i>

I hereby claim the benefit under Title 35, United States Code, § 120 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code § 112, I acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, § 1.56 which occurred between the filing date of the prior application and the national or PCT international filing date of this application.

<i>Application Number(s)</i>	<i>Filing Date (MM/DD/YYYY)</i>	<i>Status: Patented, Pending, Abandoned</i>

Declaration for Patent Application (cont'd.)Atty. Docket No. WAS 0394 PUS

I hereby appoint the following registered practitioners to prosecute this application and to transact all business in the Patent and Trademark Office connected therewith:

Ernie L. Brooks, Reg. No. 26,260; James A. Kushman, Reg. No. 25,634; David R. Syrowik, Reg. No. 27,956; Mark A. Cantor, Reg. No. 30,614; Ralph M. Burton, Reg. No. 17,748; Robert C.J. Tuttle, Reg. No. 27,962; Earl J. LaFontaine, Reg. No. 30,766; Ronald M. Nabozny, Reg. No. 28,648; Thomas A. Lewry, Reg. No. 30,770; John E. Nemazi, Reg. No. 30,876; Kevin J. Heintz, Reg. No. 29,805; William G. Abbott, Reg. No. 31,936; Donald J. Harrington, Reg. No. 17,427; Timothy G. Newman, Reg. No. 34,228; Frederick M. Ritchie, Reg. No. 18,669; Robert C. Brandenburg, Reg. No. 29,048; A. Frank Duke, Reg. No. 20,937; John M. Halan, Reg. No. 35,534; Jeffrey M. Szuma, Reg. No. 35,700; James R. Ignatowski, Reg. No. 26,741; Frank A. Angileri, Reg. No. 36,733; William G. Conger, Reg. No. 31,209; Sangeeta G. Shah, Reg. No. 38,614; Christopher W. Quinn, Reg. No. 38,274; Robert C. Jones, Reg. No. 35,209; David S. Bir, Reg. No. 38,383; Konstantine J. Diamond, Reg. No. 39,657; James N. Kallis, Reg. No. 41,102; Hugo A. Delevie, Reg. No. 32,688; Ralph E. Smith, Reg. No. 35,474; Michael S. Brodbine, Reg. No. 38,392; Jeremy J. Curcuri, Reg. No. 42,454; Mark D. Chuey, Reg. No. 42,415; John J. Ignatowski, Reg. No. 36,555; Pete N. Kioussis, Reg. No. 41,117; Stephanie M. Mansfield, Reg. No. 43,773; Mark E. Stuenkel, Reg. No. 44,364; Matthew R. Mowers, Reg. No. 44,956; Raymond J. Vivacqua, Reg. No. 45,369; Lawrence G. Almeda, Reg. No. 46,151; Ginta Kukainis, Reg. No. 46,082; Seth E. Rodack, Reg. No. 45,622; James W. Proscia, Reg. No. P-47,010.

Address all correspondence and telephone calls to William G. Conger
at Brooks & Kushman P.C., 1000 Town Center, Twenty-Second Floor, Southfield, Michigan 48075, (248) 358-4400.

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Full Name of Sole or First Inventor Philipp MüllerInventor's signature  Date 09-28-00Post Office Address Same As Residence AddressResidence Einsteinstrasse 1, D-84489 Burghausen, Germany Citizenship German**Full Name of Second Joint Inventor** Frank AchenbachInventor's signature  Date 09-28-00Post Office Address Same As Residence AddressResidence Weinleiten 48, D-84359 Simbach, Germany Citizenship German**Full Name of Third Joint Inventor** Georg EberlInventor's signature  Date 09-29-00Post Office Address Same As Residence AddressResidence Elisabethstrasse 21, D-84489 Burghausen, Germany Citizenship German